



# Lithium-ion batteries recycling

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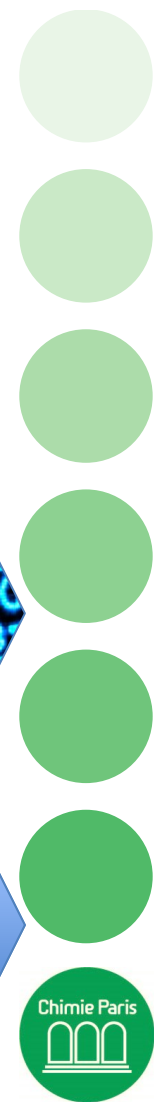
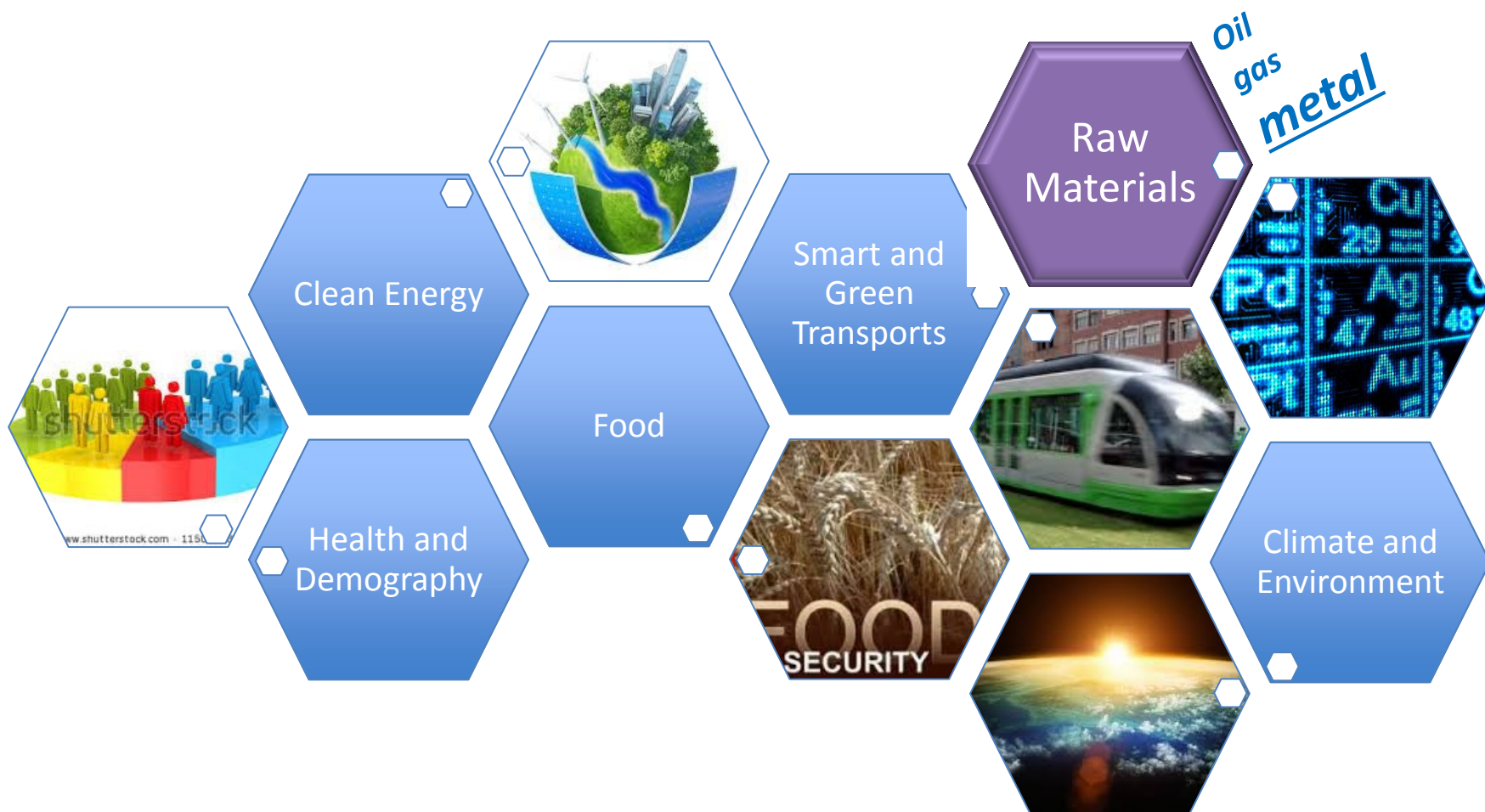
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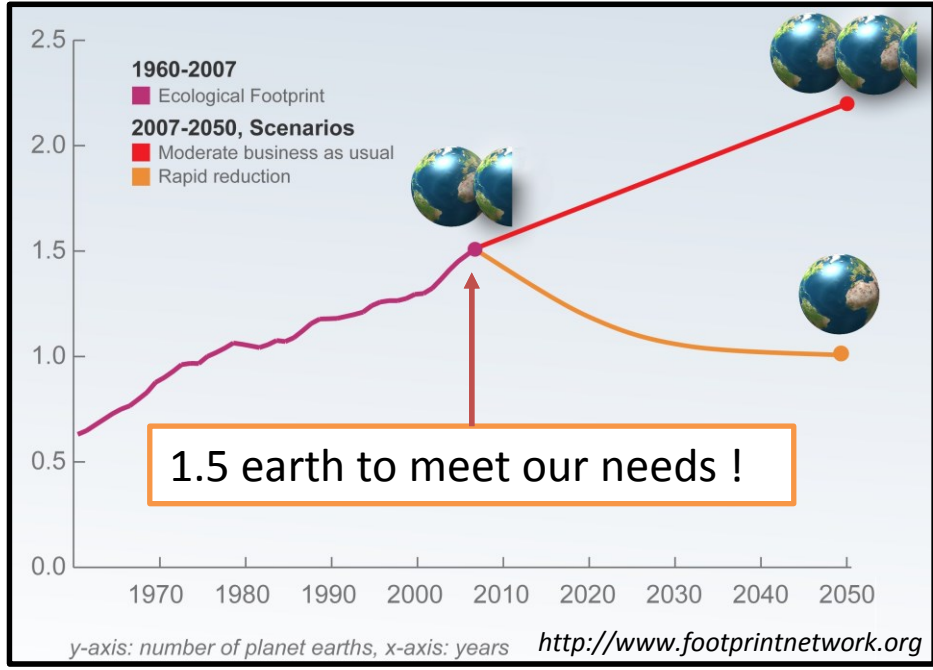


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# SOCIETAL CHALLENGES

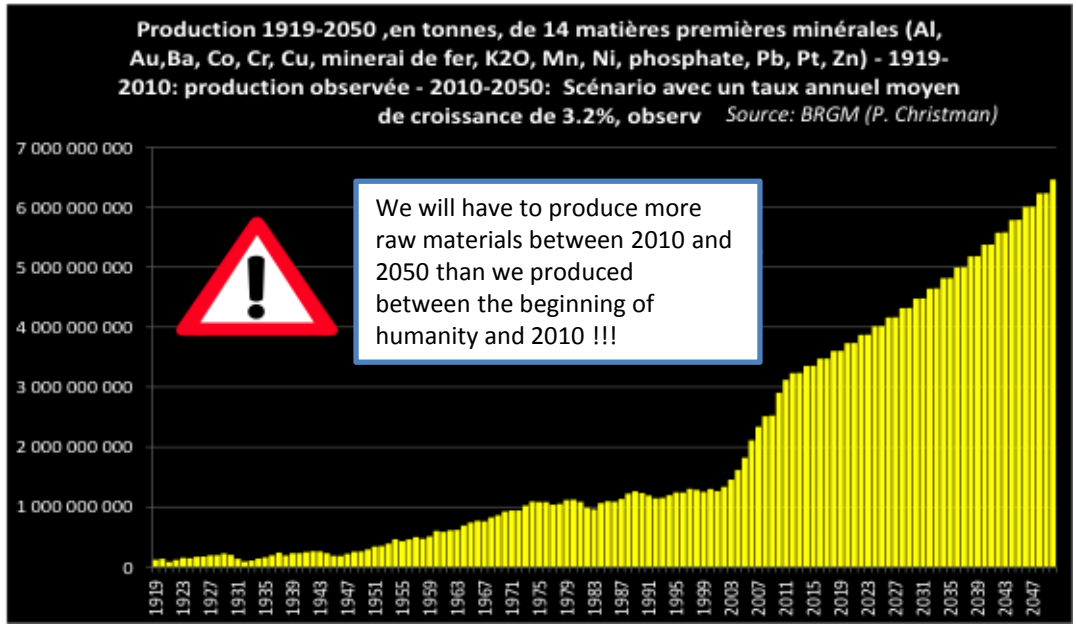
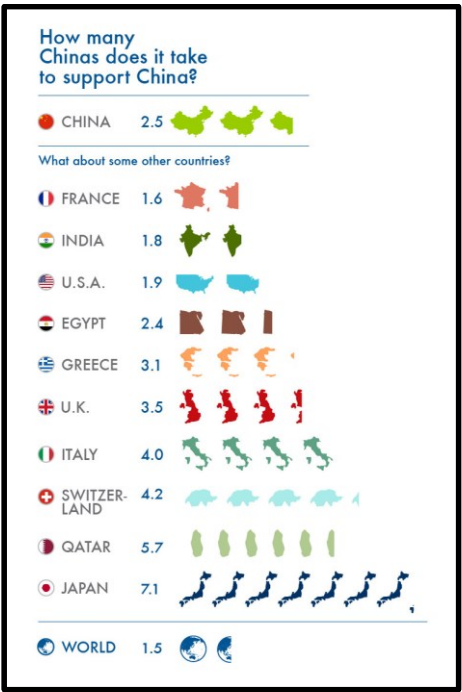




## CONSUME MORE WITH LESS

**Overshot Day** (20 August 2015) = when resource use is expected to outstrip the capacity for production—and it's getting earlier every year

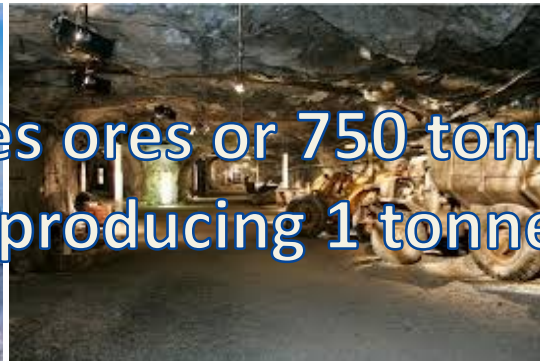
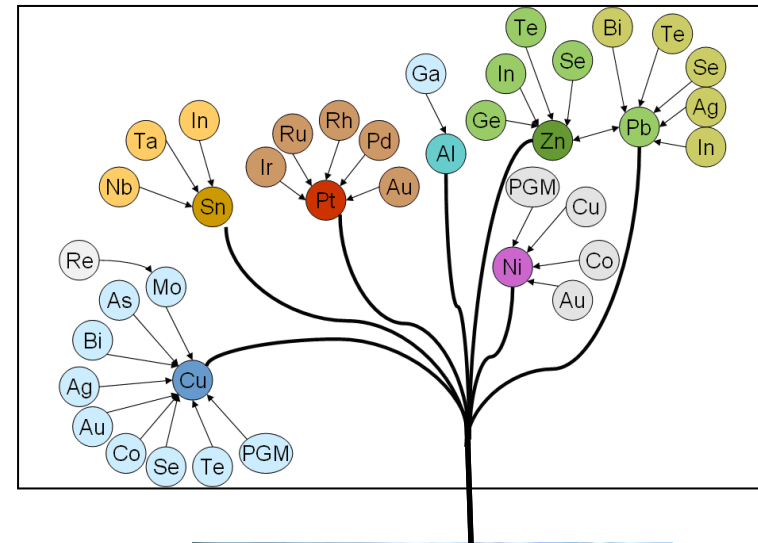
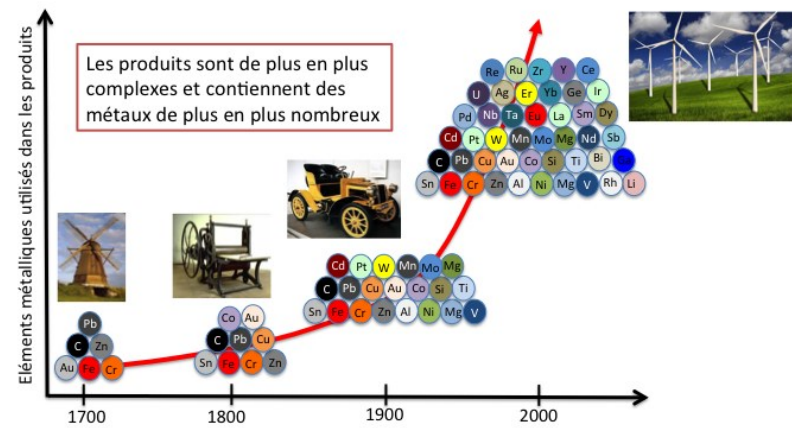
**Humanity consumed the whole resource that the earth can produce in only 8 months !!!**



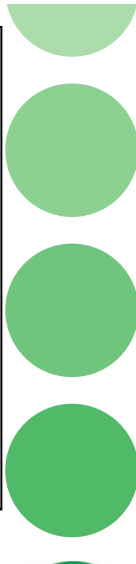


28 tonnes of Li-ion batteries  
can produce 1 tonne Li

In order to face up the world demand, we  
must extract metals (and more generally  
raw materials) contained in more and  
more complex ores from mining and  
urban mining



250 tonnes ores or 750 tonnes brine  
for producing 1 tonne Li



# We must develop new processes to recycle spent material in a sake of sustainability and to secure raw material but there are several challenges...



## Challenges in Recycle

A good opportunity to diversify the supply of raw materials

### Challenges:

- Transform small-scale industry to autonomous and profitable large-scale industry
- Develop efficient and profitable processes
- Improve the recovery of spent materials (WEEE, plastics, batteries, etc.)



## Challenges in Recycle

A resource is valuable providing that we know how to transform it!

Present processes have been developed to exploit less complex resources

Next processes should rely on:

- The complementarity of pyro and hydrometallurgy
- A fine knowledge of the resource
- The huge potential of innovation of the European actors



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Therefore, recycling appears as a good opportunity to secure raw materials. However, Secondary resources will never replace primary resources !! Each stage of the value chain of the product must be optimized...



PYRO AND HYDROMETALLURGICAL PROCESSES WERE INITIALLY DEVELOPPED FOR PRIMARY RESOURCES... THESE PROCESSES MUST BE ADAPTED TO SECONDARY RESSOURCES AND NEW COMPLEX ORES BY REDUCING PROCESSES COSTS AND ENVIRONMENTAL FOOTPRINT



# Lithium-ion batteries Recycling

1. Why to recycle?
2. How to recycle?
3. Conclusion



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# 1. Why to recycle?

## a) Economic reasons

Electric vehicle fleet is expected to significantly increase

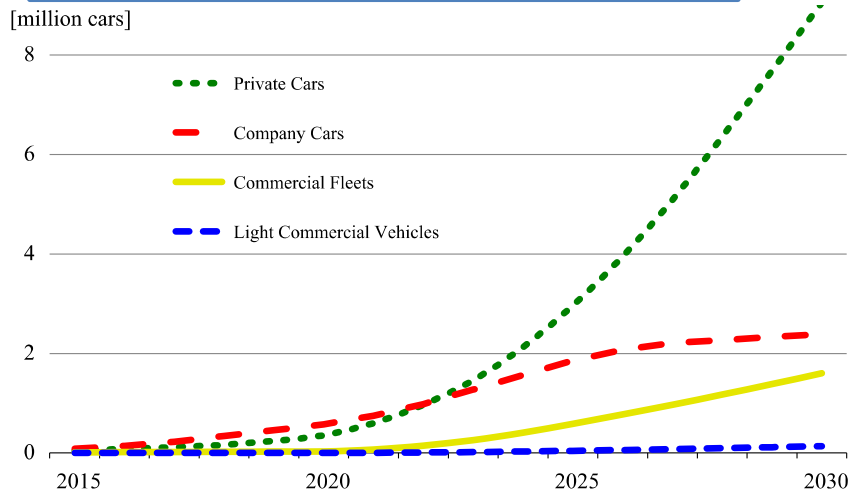


Fig. 3. Fleet size development.

Energy Policy 73 (2014) 147–157

Lithium batteries production will follow the electric vehicle production

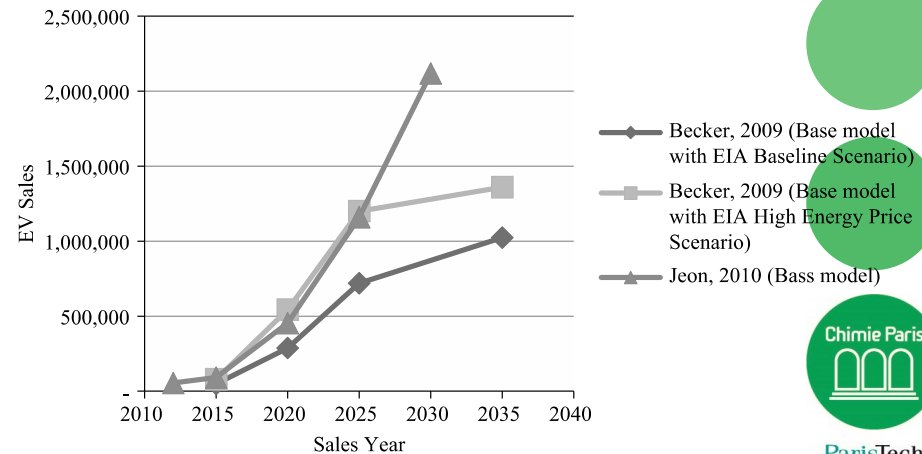


Fig. 6. EV penetration rate estimated using diffusion method [86,92].

Renewable and Sustainable Energy Reviews 21 (2013) 190–203



Components		\$/EV battery	Cost breakdown
<b>Cell Components</b>	Cathode	1,663	10%
	Anode	477	3%
	Electrolyte	447	3%
	Copper foil	184	1%
	Separator	608	4%
	Can header and terminals	1,050	6%
	Other materials	375	2%
	<b>Total material</b>	<b>4,803</b>	<b>29%</b>
<b>Cells</b>	Labor for cell manufacturing	2,586	16%
	<b>Total cell</b>	<b>7,390</b>	<b>45%</b>
<b>Electronics</b>	Mechanical components	2,053	12%
	Electrical Components	299	2%
	Electronics (battery mgmt. system)	1,381	8%
	<b>Total Electronics</b>	<b>3,733</b>	<b>22%</b>
<b>Packs</b>	Labor for pack manufacturing	268	2%
	<b>Total Packs</b>	<b>11,390</b>	<b>69%</b>
<b>Warranty</b>		228	1%
<b>Gross Profit</b>		4,979	30%
<b>Total Cost</b>		<b>16,596</b>	<b>100%</b>

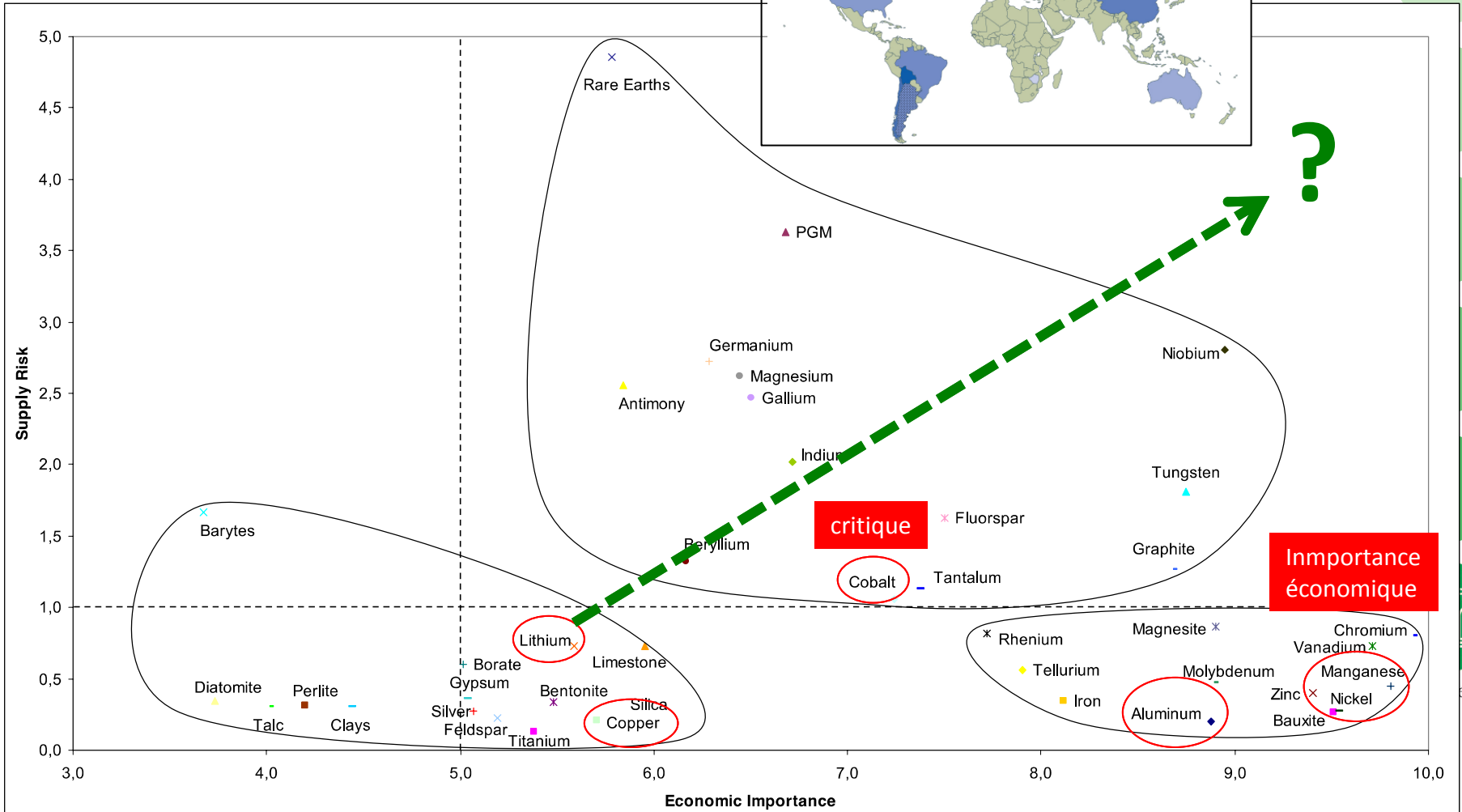
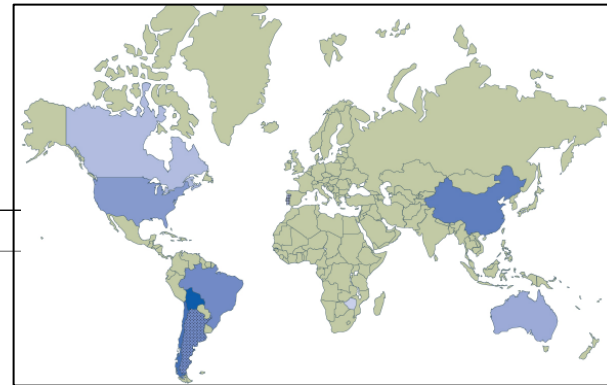
Source: Lithium-ion Batteries for Electric Vehicles: THE U.S. VALUE CHAIN (oct 2010)

Li battery recycling - April, 6th 2016, Gotteborg (Sweden)



# Where is Lithium?

## b) Geopolitical reasons



Source: commission européenne

## c) Regulations

Directive 2006/66/EC of the European Parliament and of the Council of 6 September 2006 on batteries and accumulators and waste batteries and accumulators

## d) Sustainability



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# CONTENT

1. Why to recycle?
2. How to recycle?
3. Conclusion

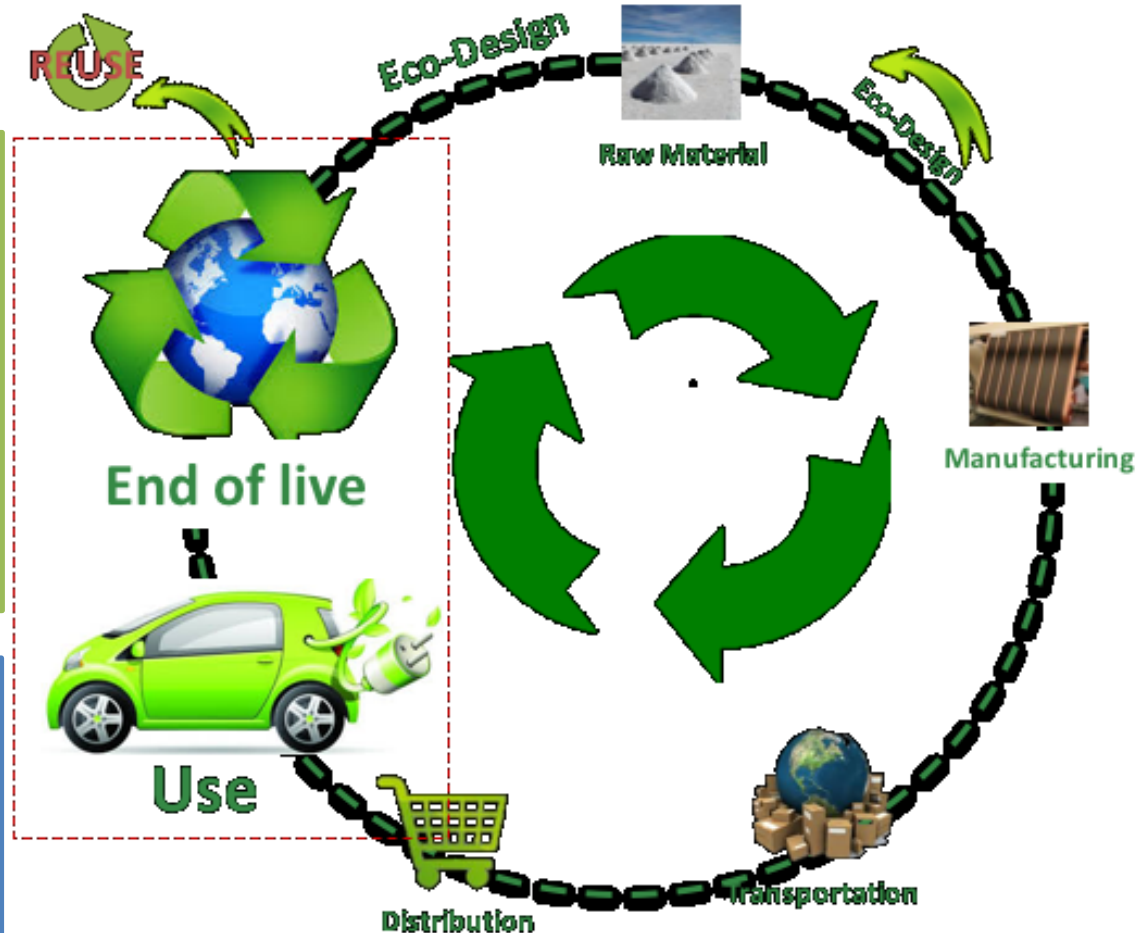


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# 2. How to recycle?

## We must think about:

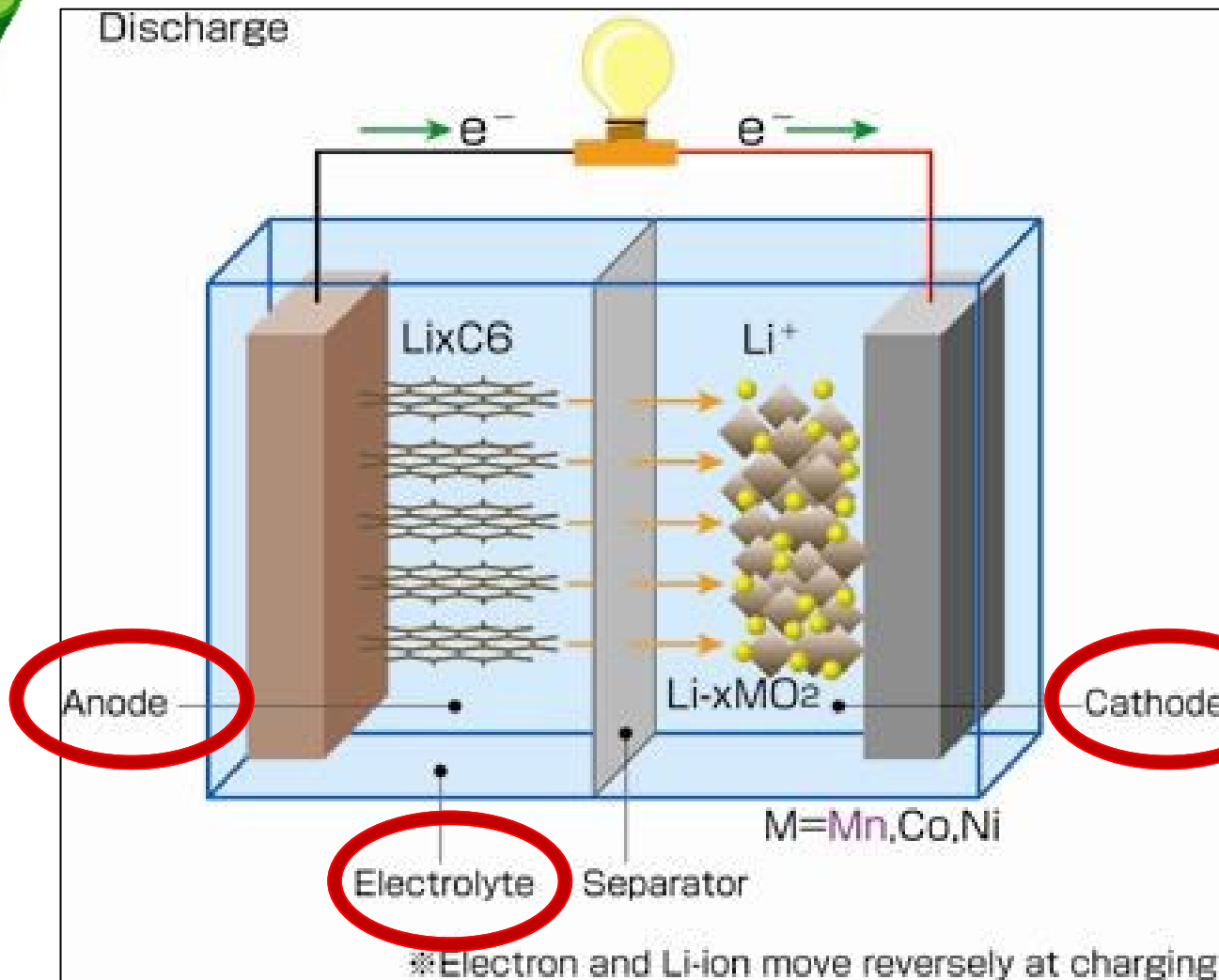
- Developing efficient, cheap and sustainable processes to recover valuable materials from spent batteries
  - These processes must be able to adapt to the large variety of technologies
- 
- Processes must be developed by thinking about the whole value chain (the process depends on what happens before and after the recycling stage)

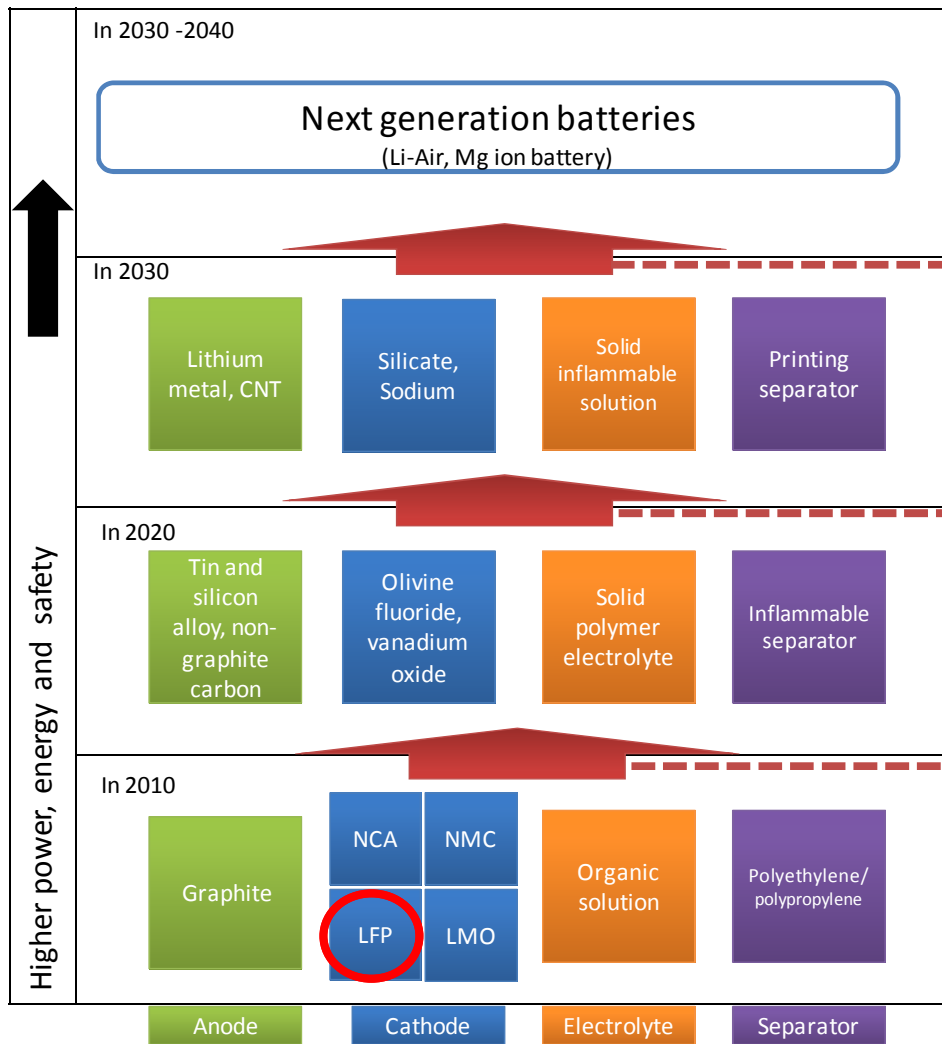




Recycling

## 2.1. Valuable components to recycle





Nanotechnology to increase battery safety and battery performance

- Nano-sized anode/cathode with higher power density
- Nano-sized anode/cathode with higher energy density
- Nano-sized anode/cathode to increase battery charging speed
- Nanomaterials to increase electric conductivity of anode and cathode (ex. CNT)
- Nanomaterials to increase thermal stability
- Nanotechnology to overcome adverse conditions
- Nanomaterial property and size control technology
- Nanomaterial evaluation technology



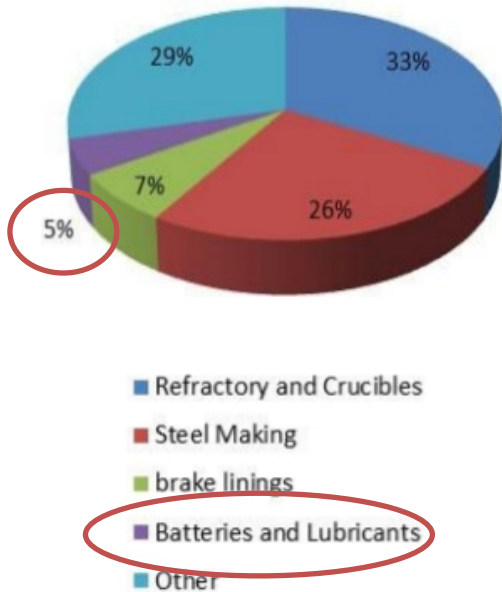
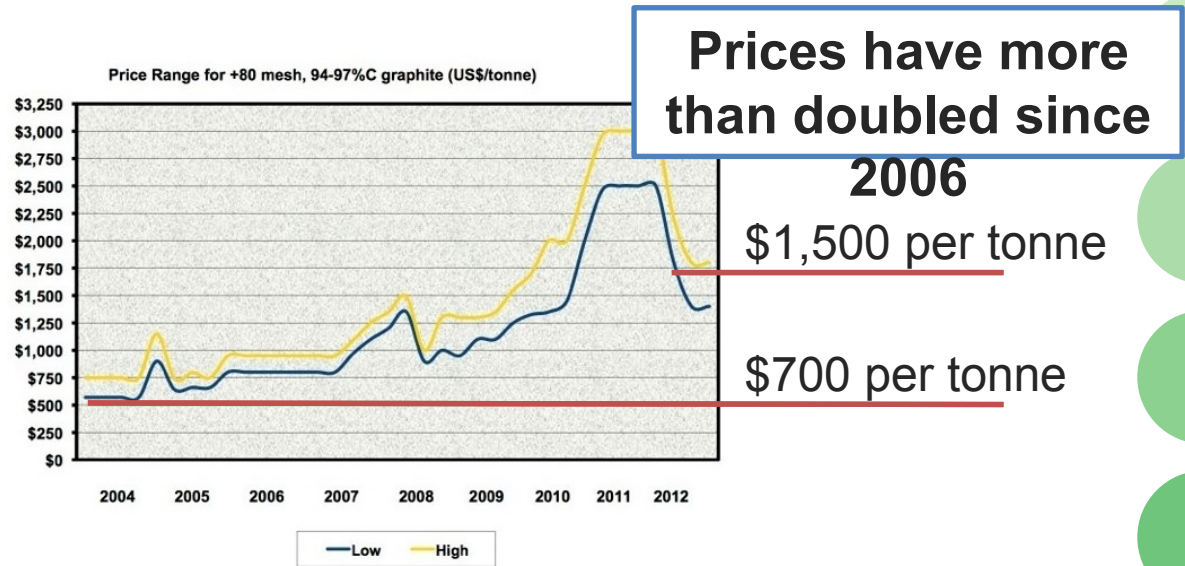
Source: Lithium-ion Batteries for Electric Vehicles: THE U.S. VALUE CHAIN (oct 2010)

*Large variety of technologies in constant evolution... the process must adapt*

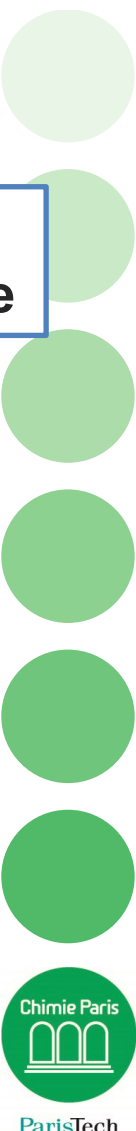




# NEGATIVE ELECTRODE = Graphite today and in the near future (cost=3%)



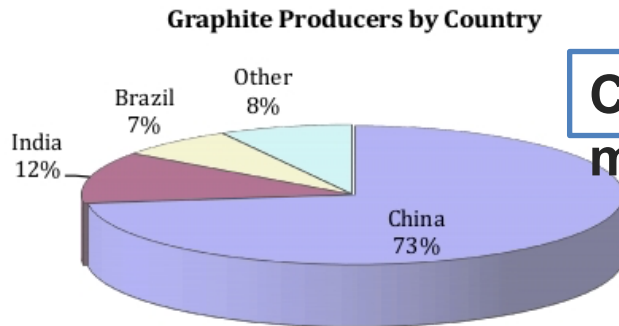
Today, 5% of graphite is used in LiBs  
 Tomorrow, this percentage will grow: 1.1 million tonnes currently being brought to market will no longer be enough to face electric vehicle market.





Recycling

# NEGATIVE ELECTRODE = Graphite today and in the near future (cost=3%)



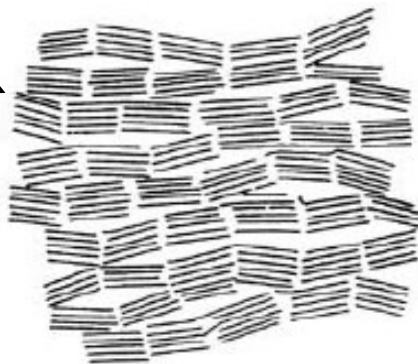
China has monopoly

- Substitute natural graphite by artificial graphite produced from natural carboneous materials
- Recycling graphite from Lithium-ion batteries

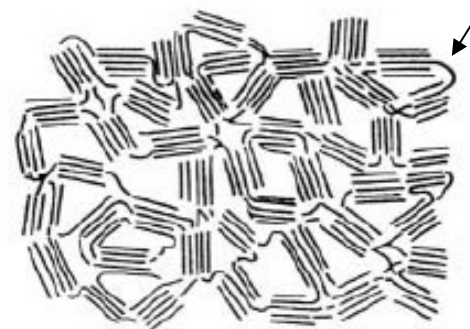
Soft carbon  
(can lead to graphite)

small crystallites  
in almost the same direction  
→ low warm → ↑ order  
→ graphitization

Turbostatic disorder



Hard carbon  
(cannot form graphite even  
at  $T > 2000^{\circ}\text{C}$ )



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Recycling

## NEGATIVE ELECTRODE = Graphite today and in the near future (Cost=3%)

There is only few researches on graphite recycling and industrial processes for LiBs recycling do not recycle graphite

We must develop efficient processes capable to recycle graphite that can be used in other applications or in lithium-ion batteries (quality of graphite?)

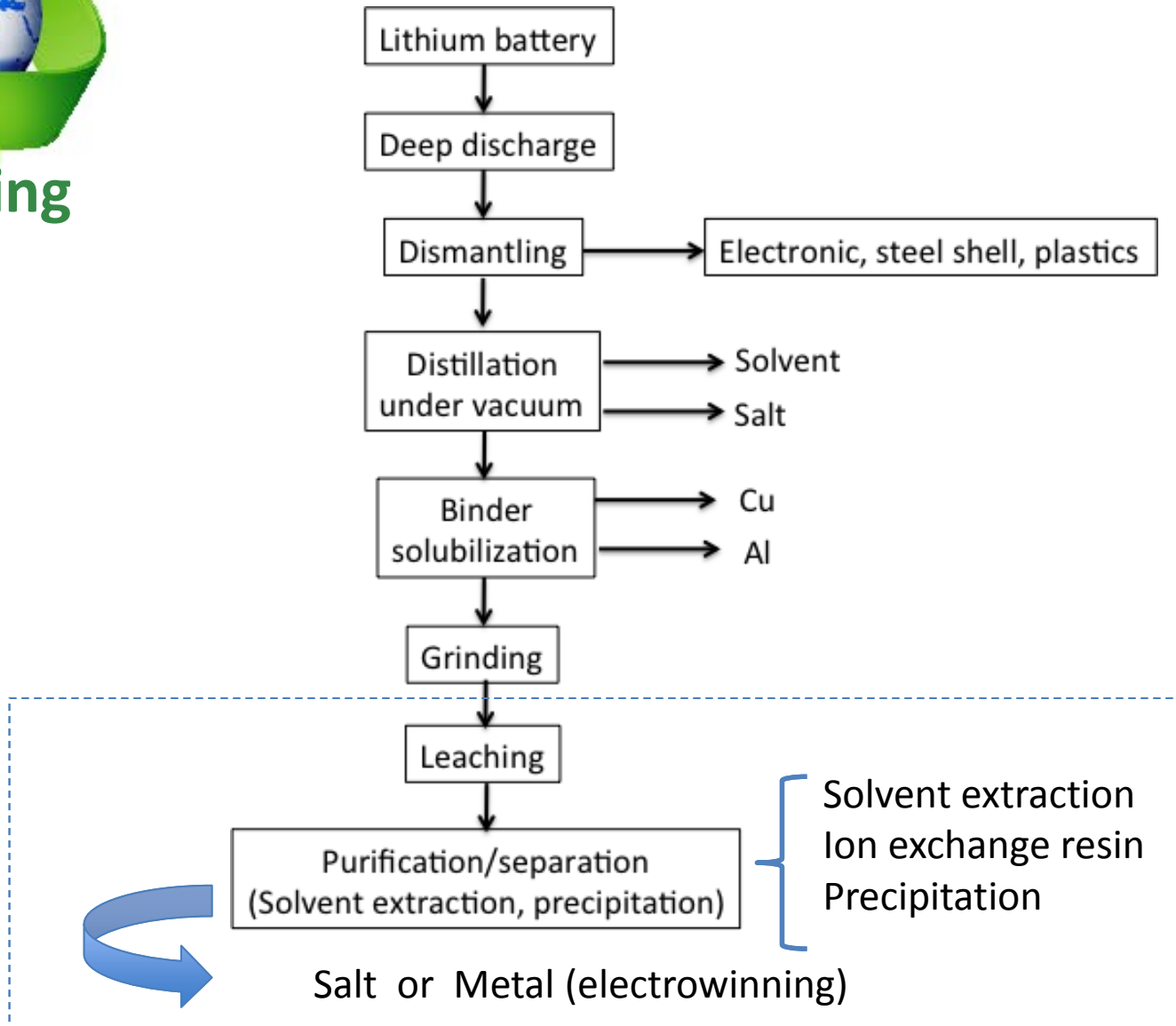
We must also develop negative electrodes based on the use of artificial graphite and increase the variety of negative electrode technologies (silicone?)



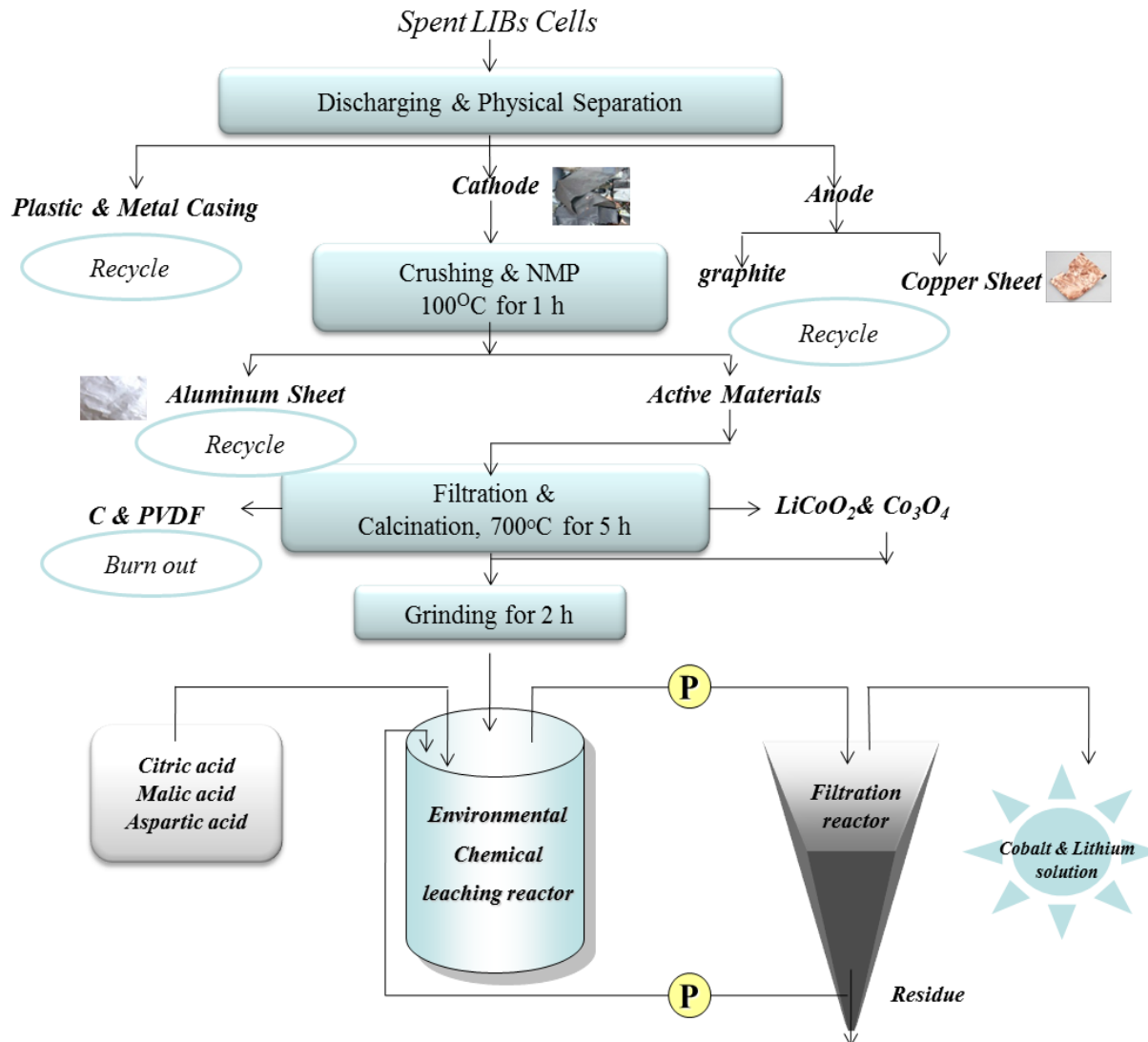
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# POSITIVE ELECTRODE (Cost=10%)



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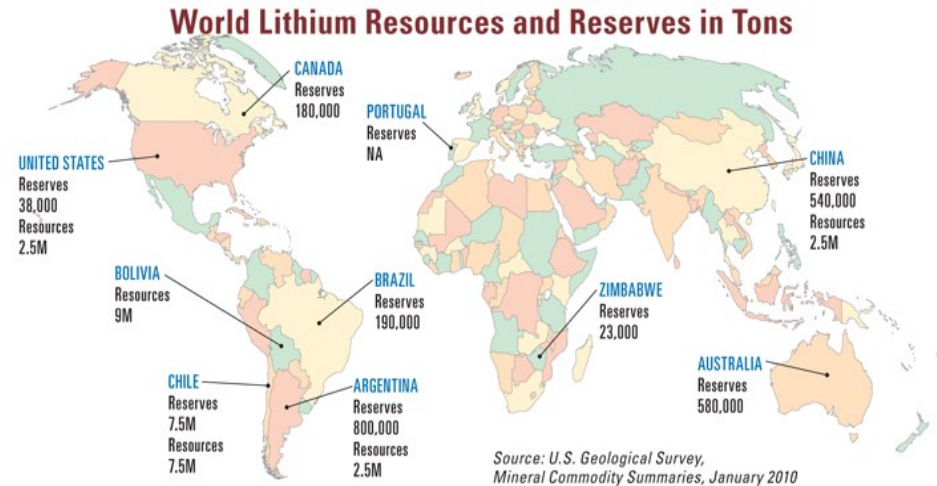
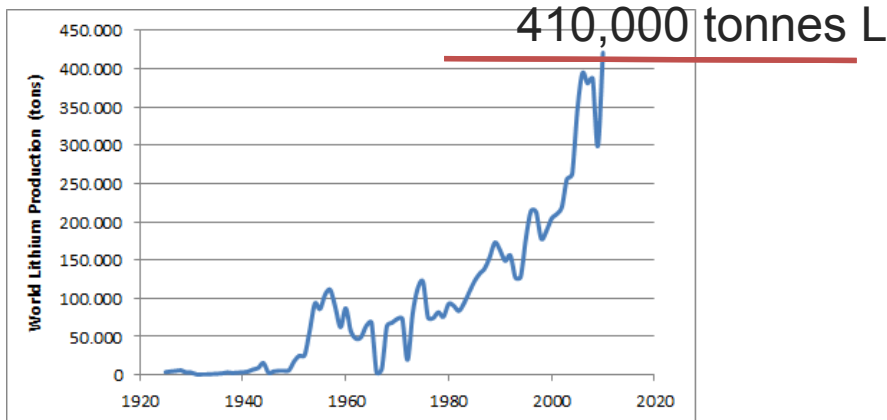


Many flow sheets based on different chemistry are available in the literature



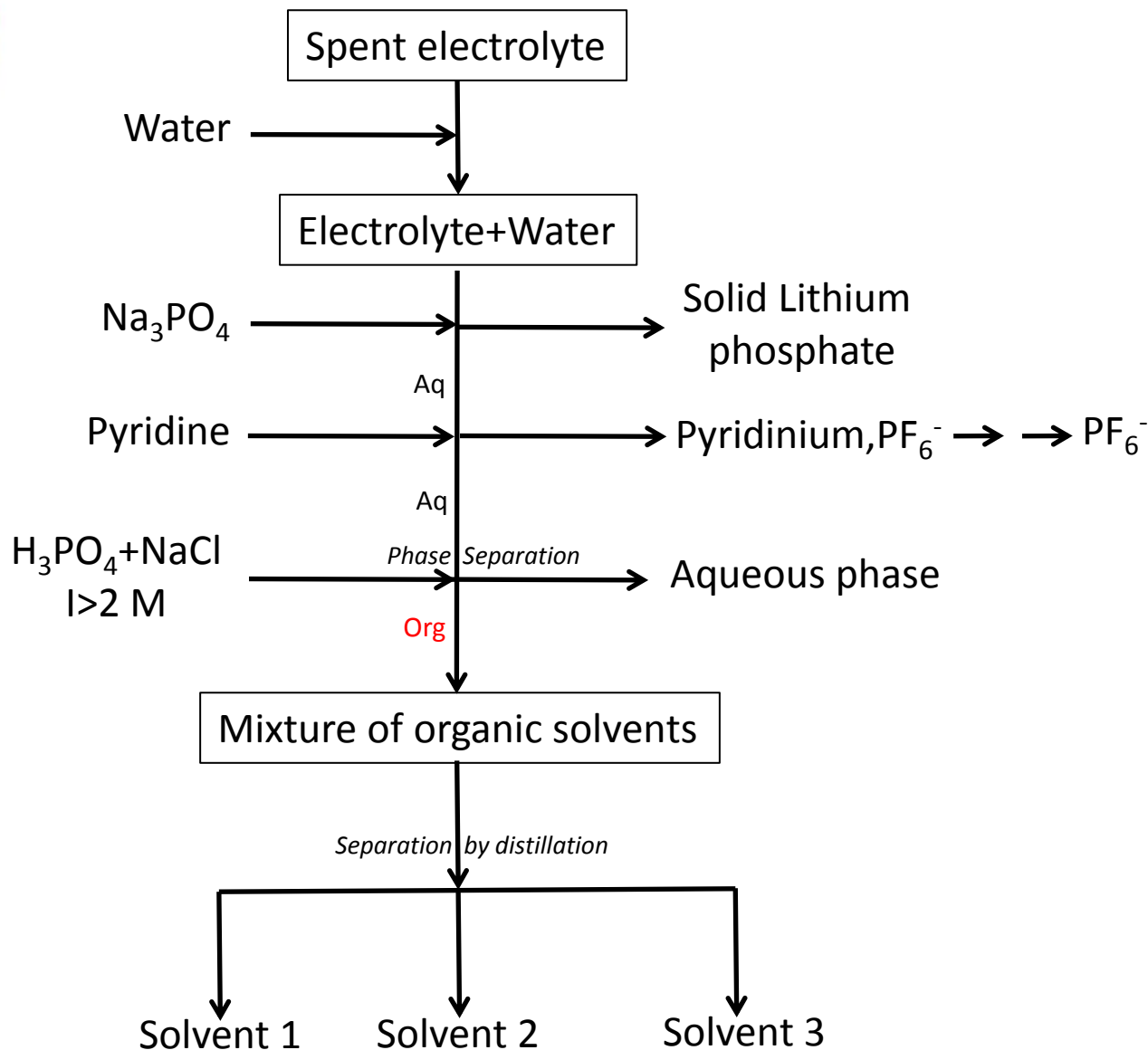
# ELECTROLYTE (Cost=3%)

Electrolyte = Lithium salts (high grade)+organic solvents



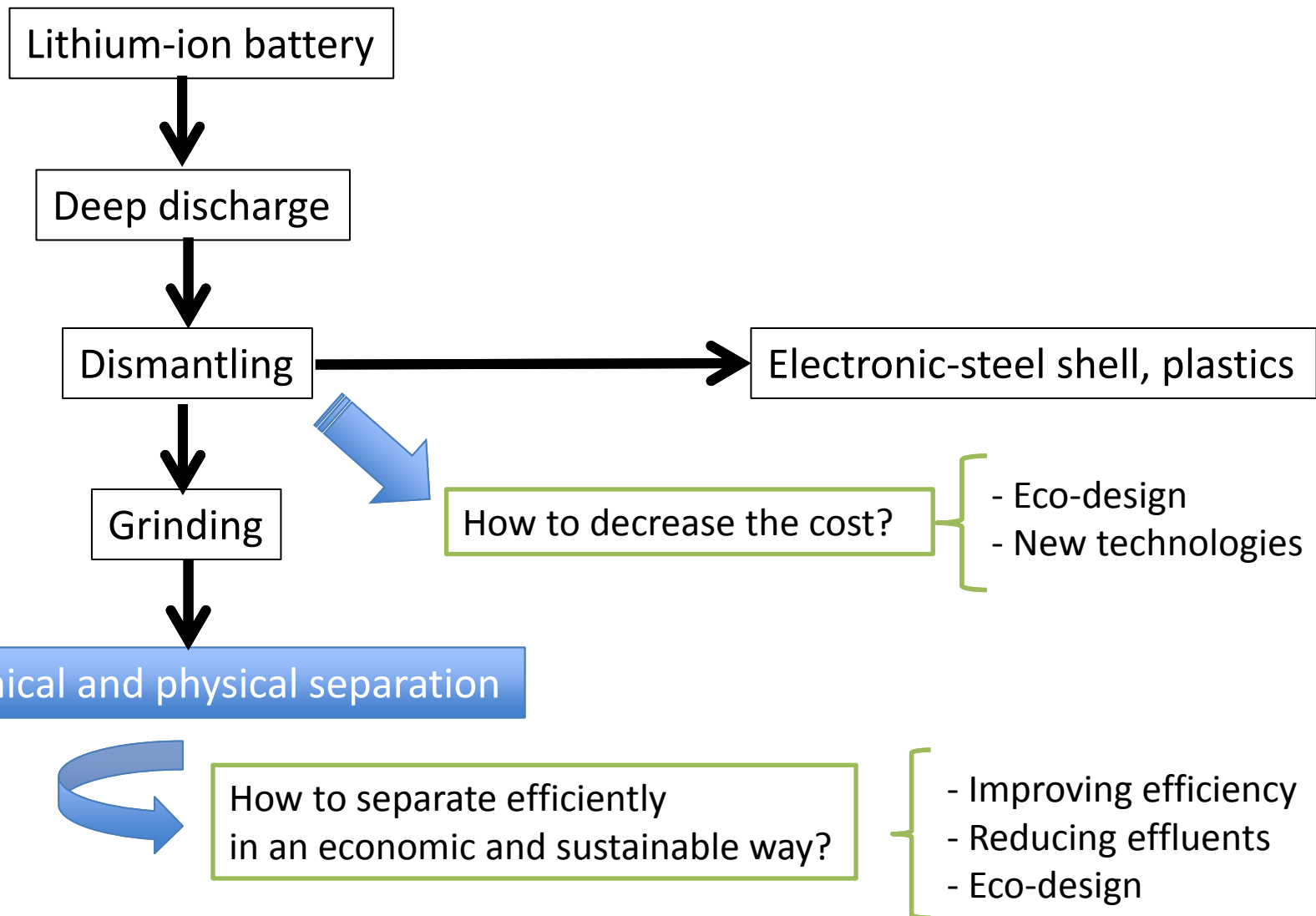


# ELECTROLYTE (Cost=3%)



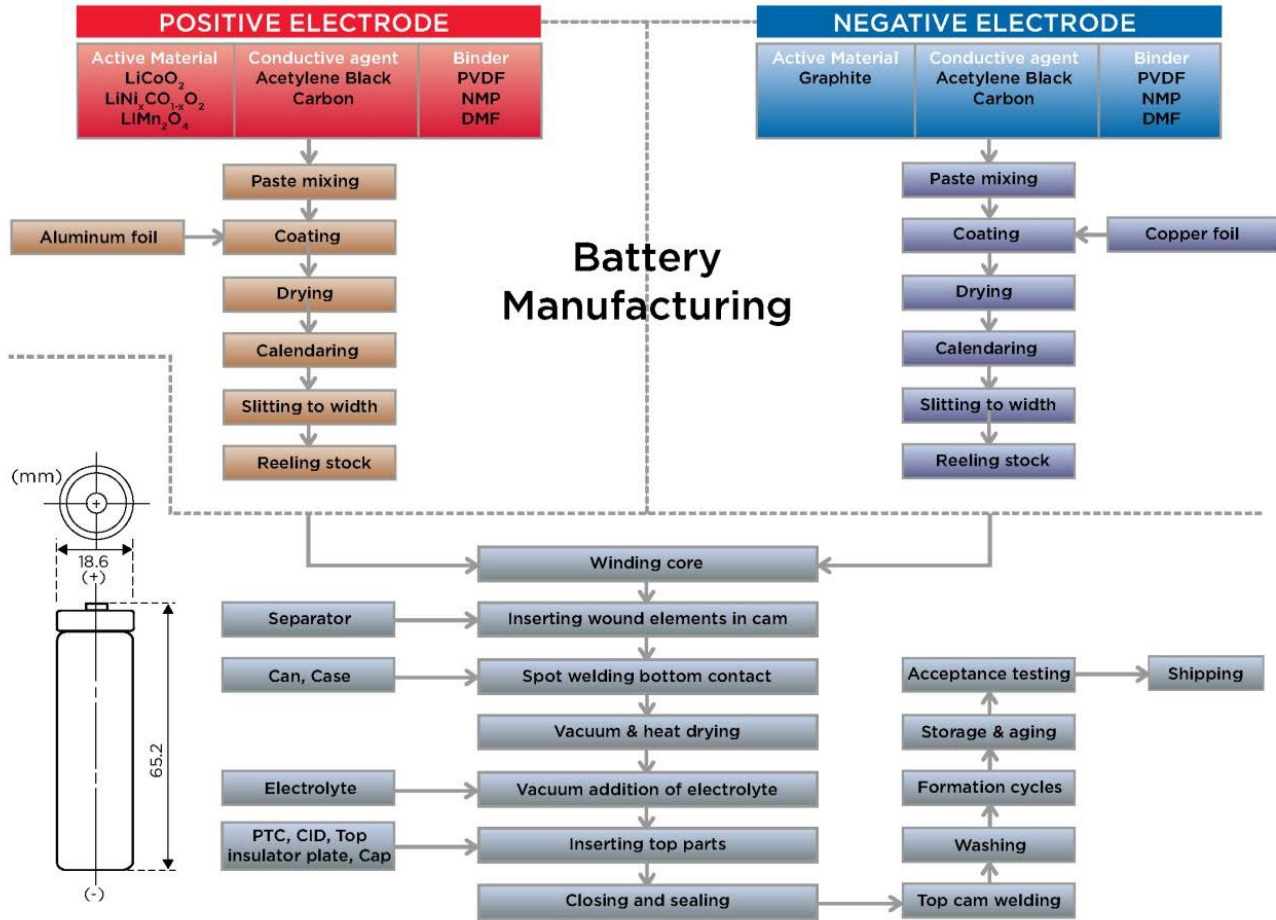
## 2.2. Recycling processes

### General Flow sheet in hydro- and pyro-metallurgy





ASSEMBLY



A

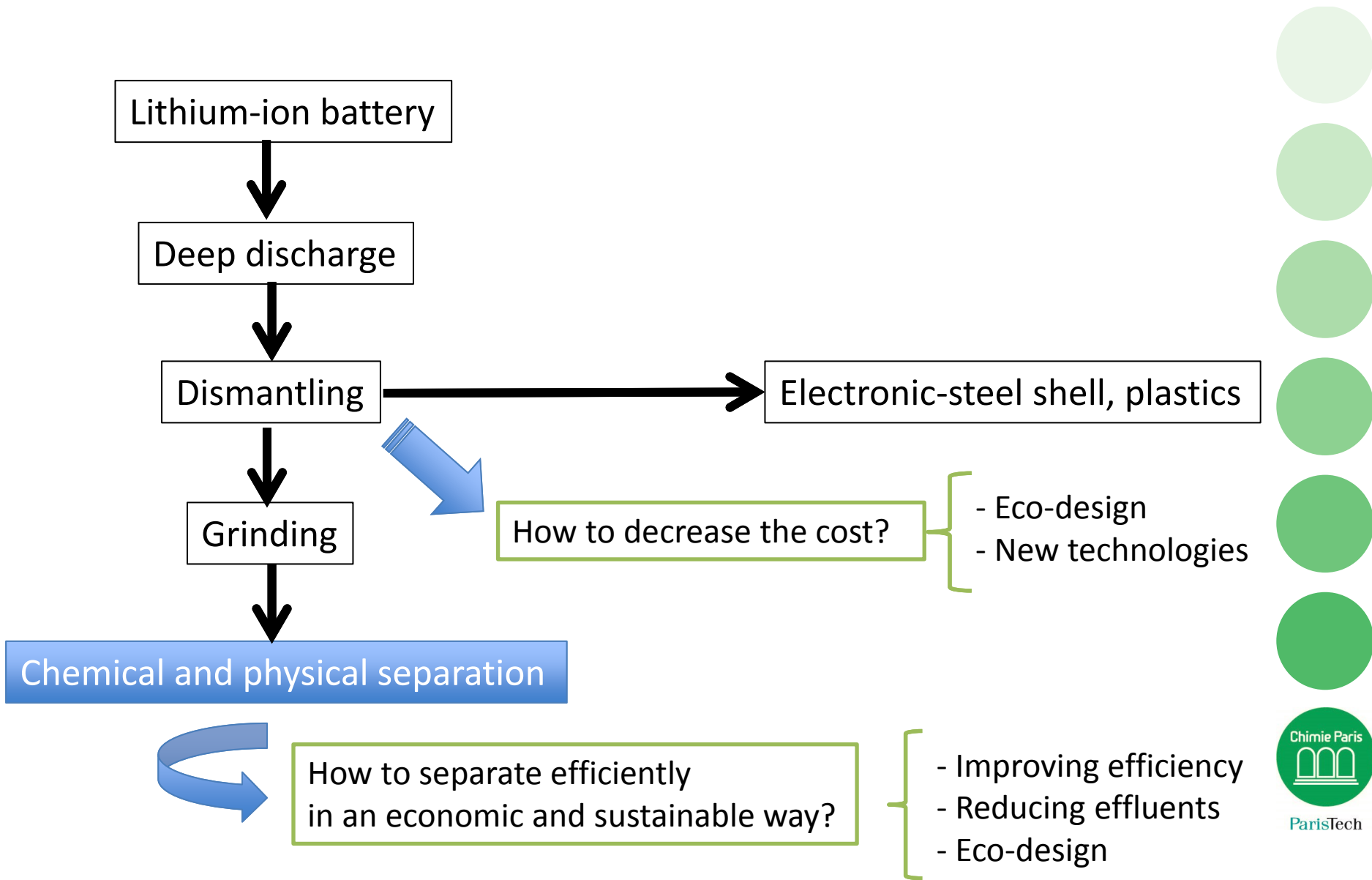
DSIMANTLING FOR

RECYCLING



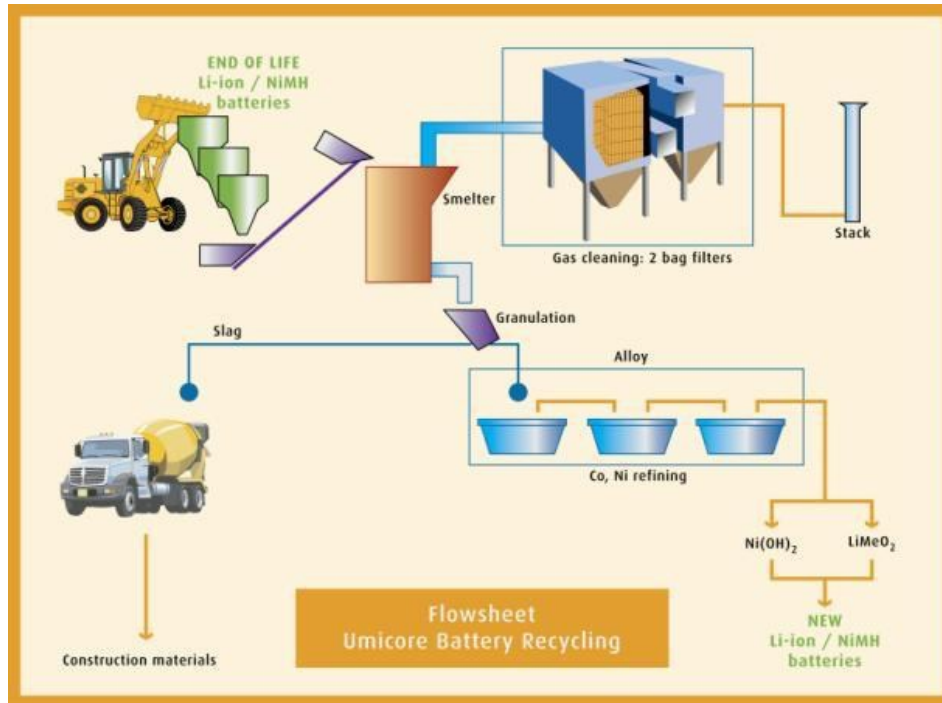
### Battery assembly process

Source: Gaines, 90th Annual Meeting of the Transportation Research Board (Janv 2011)



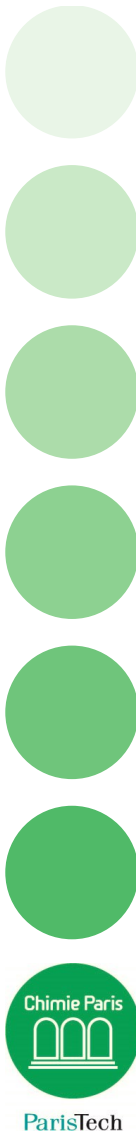
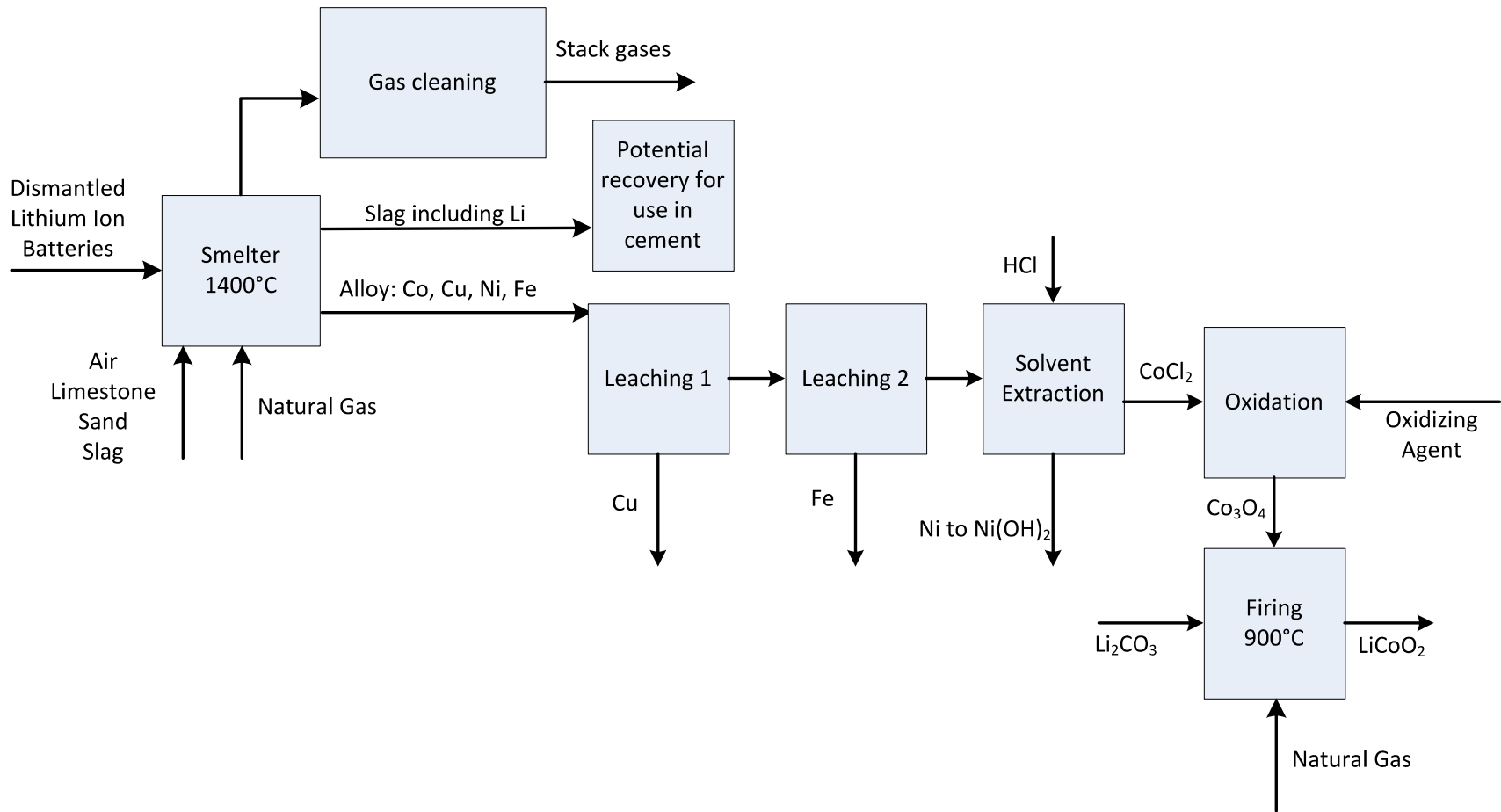
## 2.2.1. Pyrometallurgical processes for LiB recycling

### Umicore recycling process



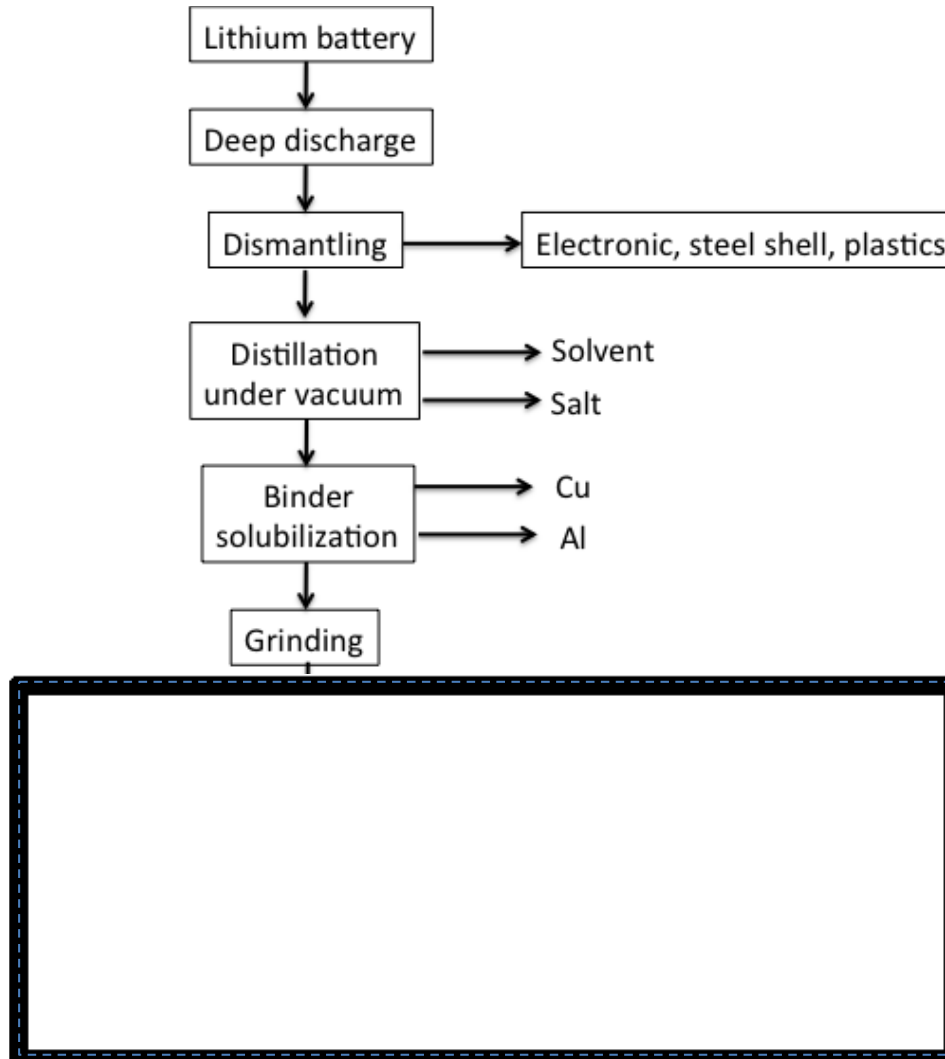
- Plastics, electrolyte solvents, and carbon electrodes are burned in the smelter (carbon = reducing agent and heat fuels the smelter).
- Co and Ni are the main products; Fe can be recovered as well
- Al and Li go to the slag for low-value uses.
- 93% recovery rate for Li-ion batteries (metals 69%, carbon 10%, plastics 15%), but a much smaller percentage actually comes out as usable high-value material.

# Detailed in the Umicore recycling process



# How to recycle?

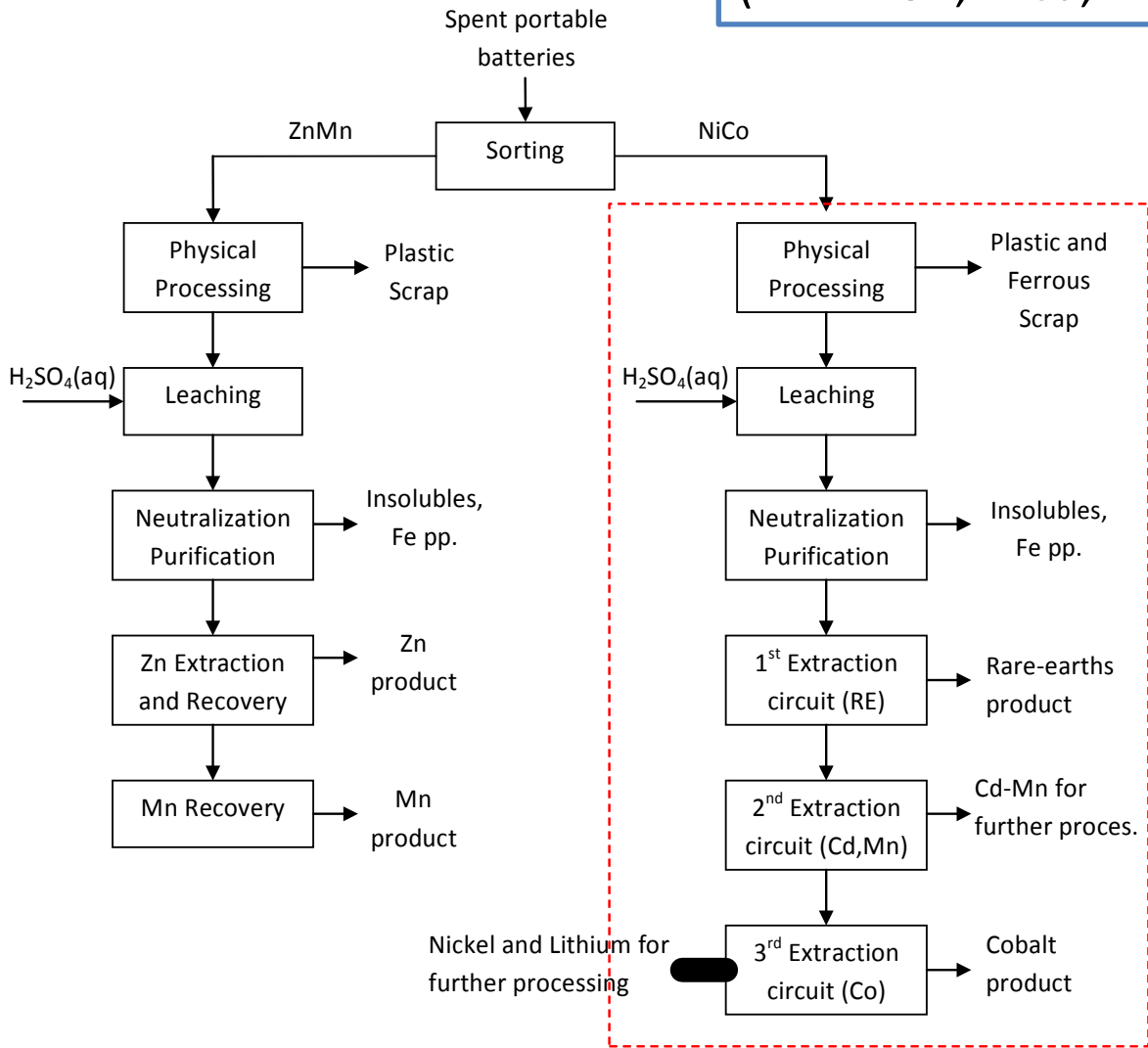
## 2.2.2. Hydrometallurgical processes for LiB recycling



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# BATMIX Process

Inlet: Mixture of different battery technologies (Zn-MnO<sub>2</sub>, NiCd, NiMH, NiCo from Li batteries)



Sieving (polymer/ferrous scrap separated from electrode materials (disaggregated=small))

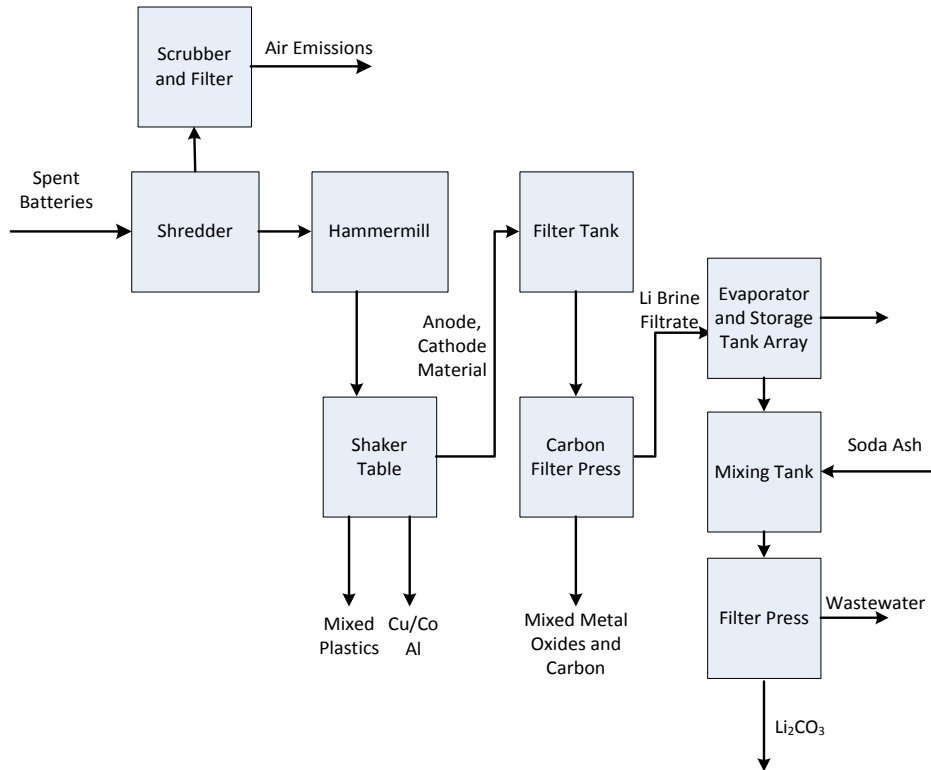
Jarosite or goethite

D2EHPA } pH  
Phase volume ratio

Cyanex 272 } pH  
Phase volume ratio

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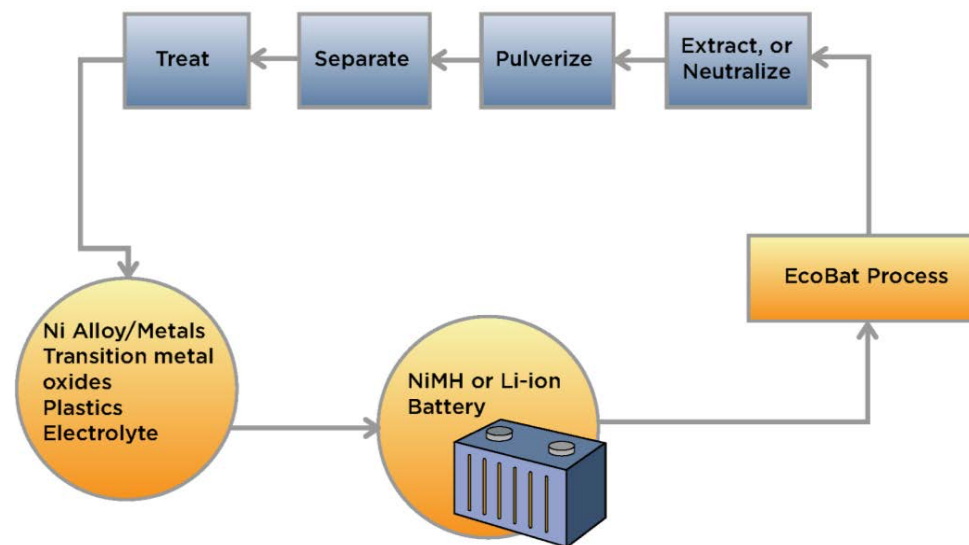
# TOXCO Process



- Recycling process is mainly mechanical and chemical  
→ Minimum emissions and energy consumption (low temperature)
- 60% of the pack materials can be recycled, and a further 10% reused.

- Battery pack is discharged for safety reasons
- Control circuits are removed and tested for possible reuse
- Wires and some other metals are removed for recycling
- Packs are disassembled
- A series of mechanical processes are used to reduce the size of the cell materials
- Products=copper cobalt and lithium carbonate

# ECOBAT Process



- Minimal energy use (no high-temperature processing)
- Many of the process details are proprietary, and so cannot be specified here.
- Electrolyte recovery using supercritical CO<sub>2</sub> for salt reuse.
- Separation processes based on surface properties and solubility.
- Over 80% of the material is actually recycled to useful products with potential value of the recovered materials quite high.
- Experimental results are excellent for both cobalt and phosphate cathodes.
- Processing a mixed feed would require additional separation steps to yield high-quality final products.
- Issue: scaling up this process.



# 3. CONCLUSION

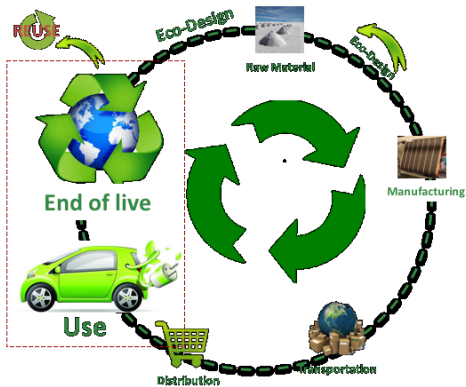
Most of the processes are pyrometallurgical processes but there is no doubt that hydrometallurgical processes will be more sustainable and efficient to produce high-grade products

However, the best compromise is to combine hydro- and pyro-metallurgical processes to achieve the best performances (pyro-pretreatment)



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CHALLENGE



## Integrated view of the chain value

*Recycling process must be viewed as a part of the chain value*

## Managing spent LiB at the end of life from the user to the mechanic



## Eco-design the materials and the batteries



Supporting the development of technologies based on the use of alternative materials instead of Li, Co, Ni and graphite based materials

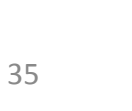




**New processes capable to process various types of streams and capable to anticipation future technologies**

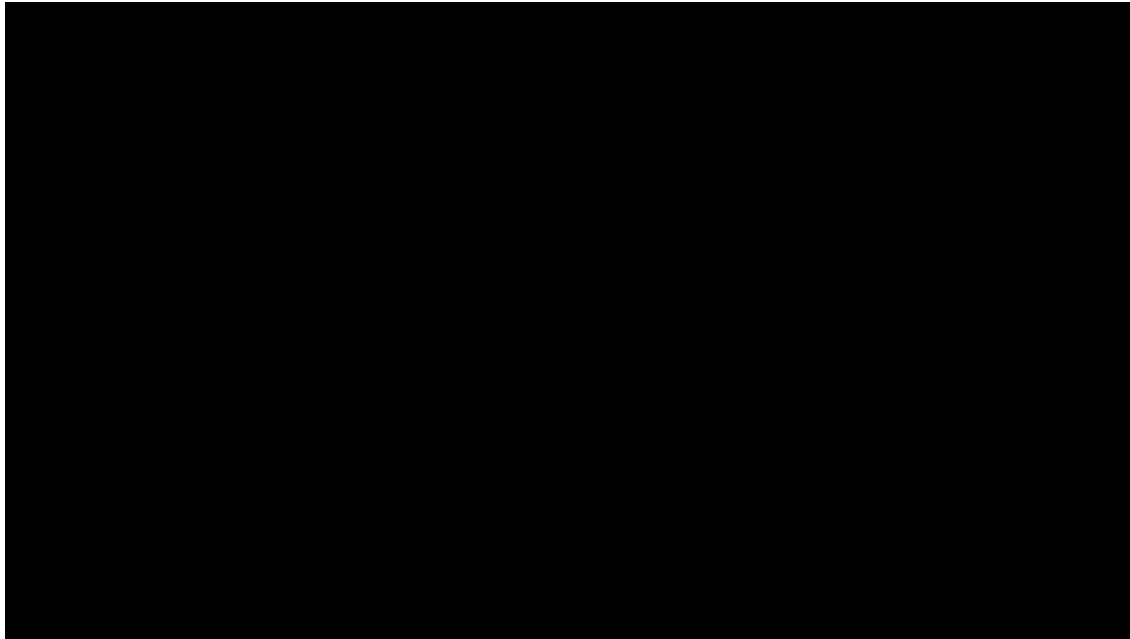


**Be able to face up the volatility of raw materials price by making innovation for reducing the recycling cost**





**Decrease recycling price by getting easier the dismantlement**



**LIAM**

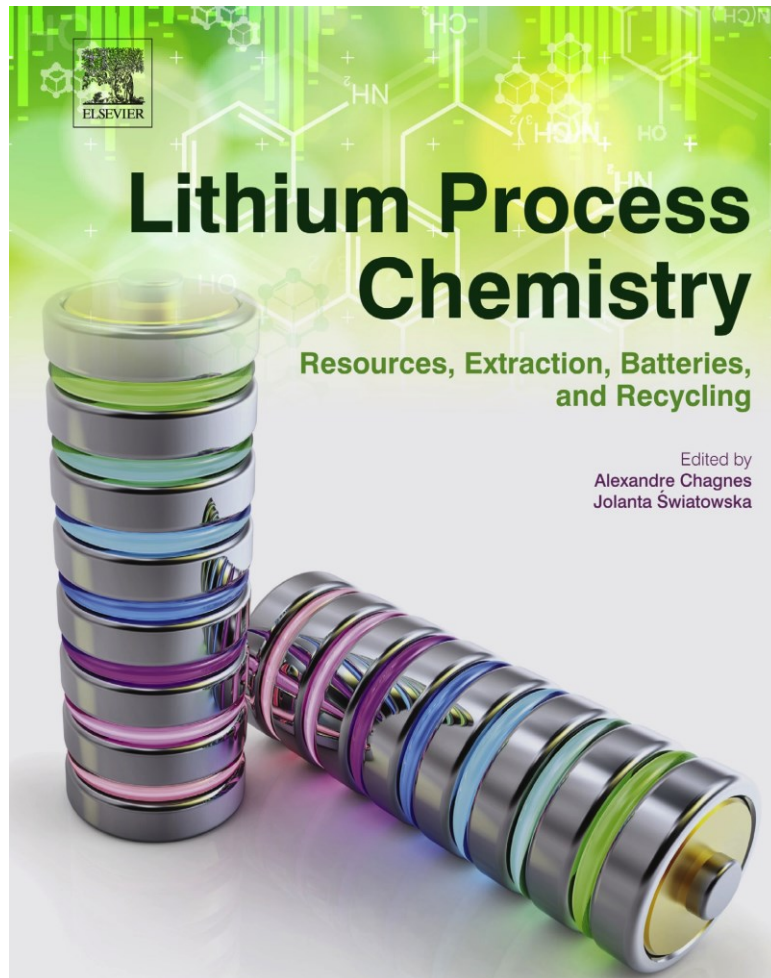


29 robots over 30 meters to disassemble 40 iPhones at the same time within 11 seconds  
→ Capacity=2.4 million iPhones a year



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If you are interested by Lithium batteries, lithium batteries recycling and e-wastes recycling...



*Published in 2015*



*Will be published in June 2016*



# Och tack för er uppmärksamhet !



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